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Chemistry Chapter 15 Test

Modern Chemistry 136 Chapter Test Name Class
Date Chapter Test A, continued _____14. A pH meter measures the pH of a solution by measuring the a. amount of charge on the hydronium ion. b. color of the solution. c. concentration of the solution. d. voltage difference between two electrodes in the meter's probe. _____15. When you conduct an ...

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Flashcards. Learn. Write. Spell. Test. PLAY.
Match. Gravity. Created by. mhardie11. Terms in this set (10) Self-ionization of water. two water molecules produce a hydroniumion and a hydroxide ion by transfer of a proton. pH.

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Modern Chemistry 141 Chapter Test Name Class
Date Chapter Test B, continued 18. When a strong acid is titrated with a weak base, the pH of the solution at the equivalence point is than 7. 19. A is a highly purified solid

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used to check the concentration of a standard solution. 20. A 1 M solution of NaOH will have a pH that is

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Modern Chemistry Chapter 15 Test Review 25 multiple choice definition of self-ionization $[H_3O^+] [OH^-] = 1 \times 10^{-14}$ definition and formula for calculating $pH = -\log[H_3O^+]$ $pH > 7$ is a base; $pH < 7$ is an acid; $pH = 7$ is neutral pH range is normally 0 to 14 calculate pH from $[H_3O^+]$ calculate $[H_3O^+]$ from pH definitions of indicators, transition interval, titration, neutralization,

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endpoint, standard solution acid-base titration monitors pH titration problems using $V_a M_a e q_a = V_b M_b e q_b$.

Modern Chemistry Chapter 15 Acid-Base Titration & pH

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Chapter 15 Test Acids And Bases Answers
Chapter Tests Acids, Bases, and Solutions (continued) 28. Dip red litmus paper into each solution and watch for color changes. Then do the same with blue litmus paper. Any solution that turns red litmus blue is basic. Any solution that turns blue litmus red is acidic.

Chapter 15 Acids Bases Test Modern Chemistry

Modern Chemistry 105 Chapter Test Name Class
Date Chapter Test A, continued Use this figure to answer questions 7 and 8. _____ 7. A solution containing 35 g of Li_2SO_4 dissolved in 100 g of water is heated from 10°C to 90°C . According to information in the figure, this temperature change would result in a. an additional 5 g of Li_2SO_4 in ...

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Modern Chemistry 68 Chapter Test Name Class
Date Chapter Test A, continued _____21. Metal
X replaces the ions of metal Y from solution,
but it cannot replace the ions of metal Z
from solution. The order these metals should
have in the activity series (from top to
bottom) is a. x, y, z b. z, x, y c. y, x, z
d. z, y, x _____22.

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Modern Chemistry 101 Chapter Test Name Class
Date Chapter Test B, continued 18. The
pressure exerted by each gas in a mixture is
called the of that gas. 19. If the
temperature and number of moles of a gas
remain constant but the volume increases, the
pressure of the gas will . 20. The lowest
possible temperature, corresponding to zero
on the ...

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highly-visible, and mission-critical

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